Basics and Theory of Electrophoresis

Jeff Prischmann
Diagnostic Lab Manager
North Dakota State Seed Department

Separation Science has become a very important tool for diagnostic and clinical applications.

Separation scientists work in a variety of areas including: biochemistry, biotechnology, food science, clinical science, etc.
Basics and Theory of Electrophoresis

- Separation scientists deal with different sample types and sample matrices and require different techniques to separate and quantify these samples.

Basics and Theory of Electrophoresis

- The two most commonly used methods of sample separation are chromatography and electrophoresis.
- Chromatography: ideal for small molecules and has high selectivity.
- Electrophoresis: most often used for proteins and other small biological molecules like DNA.

Basic Principles

What is Electrophoresis?

- A technique whereby charged molecules are separated by the use of an electric field.
- During electrophoresis, charged molecules will migrate towards an opposite charge.
Basic Principles

What is Electrophoresis?

- A mixture of molecules of various sizes will migrate at different velocities and will be separated.
- Electrophoresis is usually carried out in an aqueous solution.
- Commonly used in many areas.

History of Electrophoresis

- 1939 Zone electrophoresis developed
- 1950 Agar gel electrophoresis
- 1955 Starch gel (Smithies)
- 1957 Cellulose acetate (Kohn)
- 1959 Acrylamide gels first used (Raymond and Winstauba)
- 1961 IEF (Svensson)
- 1964 Disc Gel Electrophoresis (Ornstein and Davis)
History of Electrophoresis

- 1969 SDS electrophoresis (Beber and Osborn)
- 1971 SDS electrophoresis (Laemmli)
- 1971 Cellulose acetate gels (Meera)
- 1975 2-D electrophoresis
- 1977 Sequencing gels first used
- 1979 Agarose gel electrophoresis
- 1983 Pulsed field electrophoresis
- 1983 Capillary electrophoresis

Types of Electrophoresis

- Free solution electrophoresis (Capillary Electrophoresis)
- Supporting medium electrophoresis (paper, film, various gels)

Gel Electrophoresis
Gel Electrophoresis

- Electrophoresis that involves the use of a gelatinous material such as agarose, acrylamide, starch or cellulose acetate as the matrix.
- The gel acts as a support medium for the sample.
- Commonly used to separate samples containing proteins or DNA.

Gel Electrophoresis

- An important purpose of a gel matrix is to introduce a sieving action which allows separations of molecules based on molecular size.
- Gel matrix viscosity, density, and pore size are all factors in determining the 'speed' of separation.

Gel Electrophoresis

- 2 Main Types of Gels
  - Slab gels
  - Tube gels
**Gel Electrophoresis**

Types of Separation

- Native: separation by size and charge (charge/mass)
- Denaturing: separation by size
- Others (IEF, 2-D)

**Gel Electrophoresis**

Native

- continuous system—gel and tank buffers are the same, single phase gel; examples are PAGE, agarose, and starch gels.
- discontinuous system—gel and tank buffers are different, two-phase gel (stacking gel); example is PAGE.

**Gel Electrophoresis**

Denaturing

- SDS (sodium dodecyl sulphate) used to denature proteins (discontinuous system).
- urea or formamide used to denature DNA or RNA.
**Gel Electrophoresis**

Other types

- Isoelectric focusing: protein separation based on isoelectric points in a pH gradient.
- 2-D electrophoresis: combination of IEF and SDS-PAGE.

**Gel Electrophoresis**

Gel Types

- Starch
- Acrylamide
- Agarose
- Cellulose acetate
- IEF

**Gel Electrophoresis**

Main factors that affect separation:

- Resistance (pore size)
- Buffer strength
- Gel Temperature
- Sample
- Gel type
Gel Electrophoresis

Resistance-Pore Size

Gel electrophoresis units are simple DC circuits.

Source: National Diagnostics

Gel Electrophoresis

Resistance-Pore Size

Source: National Diagnostics

Gel Electrophoresis

Buffers

- Common buffer components include: acetic acid, boric acid, citric acid, glycine, phosphoric acid, tris, tricine.
- Ionic strength of buffers are important.
- Homogenous buffers vs multiphasic buffers
Temperature management is critical to achieve good results. Some applications require maintaining a high temperature (denaturing PAGE of DNA/RNA). Other applications require a cool temperature to prevent sample degradation or gel melting.

### Sample Types

- Proteins
- Nucleic Acids
Proteins

- Comprised of amino acids
- Protein structure is described as primary, secondary, tertiary, and quaternary based on amino acid sequence, bends/spirals, shape, and multiple peptide chains
- A protein's structure affects how it is separated during electrophoresis

Sample Types

- Types of Proteins
  - Enzymes (albumins)
  - Regulatory proteins
  - Transport proteins
  - Storage proteins (prolamins, globulins)
  - Structural proteins (glutelins)

Sample Types

Proteins - Amino Acids
Aspartic acid and glutamic acid impart acidic properties to proteins.

Lysine, Arginine, and Histidine impart basic properties to proteins.

Proteins are charged molecules and at neutral pH are either basic or acidic depending upon their AA composition.

Most proteins placed into basic conditions become negatively charged. Acidic conditions cause most proteins to develop a positive charge.
Proteins and Electrophoresis

- Native electrophoresis of proteins generally occurs in basic conditions.
- Denaturing protein electrophoresis causes proteins to become negatively charged (SDS-PAGE).
- Acidic running conditions can also be used.

Sample Types

- Isozymes (isoenzymes): enzymes that differ in amino acid sequence but catalyze the same chemical reaction.
- Isozymes have different molecular weights and chemical properties and thus can be separated from each other during electrophoresis.
- Isozymes are coded by genes on different loci.

Sample Types

- Allozymes: a variant form of an enzyme that is coded for by different alleles at the same locus.
- Allozymes also have different molecular weights and chemical properties and can be separated using electrophoresis.
Proteins and Electrophoresis

- **Monomeric enzymes**: consists of a single polypeptide unit, each band represents a different polypeptide coded by a different allele.
- **Dimeric enzymes**: quaternary structure consisting of 2 polypeptide units, different gel bands represent different combinations of polypeptides.

---

**Proteins and Electrophoresis**

In Summary: The electrophoretic mobility of a protein is dependent upon its charge, size, and shape. The type of electrophoresis that is used affects a protein’s separation.

---

*Source: Allozyme Electrophoresis and Population Structure in the Snowy Campion (Silene latifolia ssp. Alba). Vanderbilt University. (www.cas.vanderbilt.edu/bsc111b/index.htm).*
**Sample Types**

Nucleic Acids and Electrophoresis

- DNA is negatively charged (due to phosphate backbone).
- Electrophoresis of double stranded DNA occurs under native gel conditions.
- Electrophoresis of single stranded DNA occurs under denaturing conditions (ensures single stranded molecules and prevents base pairing).

---

**Sample Types**

Nucleotide Structure

![DNA and RNA structures](image)

Source: National Diagnostics

---

**Sample Types**

RNA Structure

![RNA structure](image)

Source: National Diagnostics
Sample Types
DNA Structure

Source: J. L. Schumann

Gel Electrophoresis
Main Steps
- Prepare samples
- Prepare gel and buffers
- Load samples onto gel
- Run gel
- Stain gel
- Interpret/analysis of gel
- Archive (photograph, dry gel)

Gel Electrophoresis
Results
- Electrophoretogram: The results of an electrophoresis test.
- Zymogram: banding pattern of enzymes or isozymes after electrophoresis.
Gel Electrophoresis

Results

- Unique bands and banding patterns.
- Molecular weight standards can be run with samples to estimate sizes.
- Band naming systems have been developed for some types of electrophoresis.
- Migration or separation of a sample can be measured (Rf values).

Gel Electrophoresis

Power supply
Cooling Apparatus
Electrophoresis gel apparatus
White Light/UV Light Box
Digital Camera/Gel Documentation System
Reagents: Gel staining chemicals, pre-made gel or gel chemicals, buffers, etc.
General lab equipment: pH meter, pipettors, scale, stir plates, etc.

Power Supplies

- Power supplies vary based on type of electrophoresis and type of gel.
- PAGE usually uses higher voltage.
- Agarose gel electrophoresis generally uses lower voltage.
- Some applications may use either constant current, constant watts, or constant volts.

Equipment

- Constant voltage generally used for SDS-PAGE of proteins.
- Constant power or constant current used for many types of denaturing PAGE.
- Effective way to manage heat during electrophoresis is to use constant current or constant power.

Equipment

Electrophoresis Gel Apparatus

- Vertical apparatus: used for PAGE
- Horizontal apparatus: used for agarose gels, starch gels, IEF.
Equipment

Gel Electrophoresis Apparatus Manufacturers

- C.B.S. Scientific (www.cbsscientific.com)
- Bio-Rad (www.bio-rad.com)
- Hoefer (www.hoeferinc.com)
- Owl Separation Systems (www.owlsci.com)
Equipment
Digital Camera/Gel Documentation System

Applications
- DNA Sequencing
- Blotting (DNA Hybridization /Southern Blot)
- Medical Research
- Protein research/purification
- Agricultural testing
- Many others

Applications
- Seed Testing
  - Variety Identification
  - Varietal purity (identify mixtures, off-types, etc.)
  - GMO/Adventitious presence Testing
Applications

Seed Testing-cont.

- Measurement of hybrid purity
- Identify genetic markers used by plant breeders
- Seed production/certification (monitor varietal purity and identity)

Applications

Seed Testing-cont.

- Quality control in processing and other industries.
- Variety registration (PVP applications)
- Documentation of genetic resources in germplasm collections, etc.

References

- Westermeier, R. 2001 *Electrophoresis in Practice*.
- Chrambach, A. 1992 *Advances in Electrophoresis*.
- Hames, B. 1998 *Gel Electrophoresis of Proteins: A Practical Approach*.
References

- Electrophoresis journals: Journal of Separation Science (JSS), Electrophoresis.
- Societies: The American Electrophoresis Society (www.aesociety.org)

Questions